

Visible Light Photochemistry in Epitaxial N-doped TiO₂ Anatase

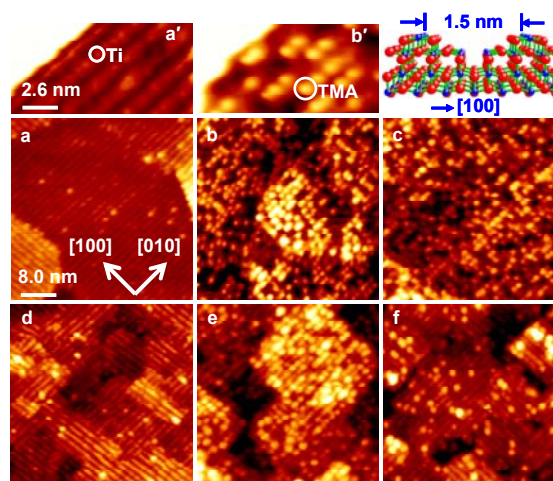
S.A. Chambers,¹ T. Ohsawa,^{1,2} Y Du,¹ M.A. Henderson,¹ V. Shutthanandan¹

¹PACIFIC NORTHWEST NATIONAL LABORATORY Richland, WA 99352 USA, ²Department of Chemistry, TOHOKU UNIVERSITY, Sendai, JAPAN.

Much effort has been devoted to heterogeneous photocatalysis using oxide semiconductors. TiO₂ has been widely used since its utility for photoelectrolysis of water was first reported [1]. TiO₂ has also attracted much attention for use in dye-sensitized solar cells [2, 3] and the photodecomposition of organic pollutants [4,5]. However, with a 3 eV bandgap, TiO₂ absorbs relatively little of the solar spectrum. Asahi *et al.* [6] first suggested that nitrogen doping reduces the bandgap. There have since been a plethora of publications describing enhanced visible-light photochemical activity in N-doped TiO₂ nanoparticles. While intriguing, these studies lack the in-depth materials understanding required to establish defensible cause-and-effect relationships between composition/structure and electronic structure/photochemical properties. Such determinations are difficult in light of the heterogeneous, multiple-orientation, and defective nature of nano-particle assemblies. Electronic structure calculations predict that N 2*p*-derived states are localized about substitutional N at O sites, and fall at midgap or close to the valence band maximum. The absence of band formation at typical doping levels ($\leq \sim 1$ at. % substitution for O) suggests that holes created by light absorption should become trapped at N sites, precluding surface photoactivity.

Indeed, we have found this to be the case for epitaxial films of N-doped rutile grown by molecular beam epitaxy (MBE) in three different orientations. In contrast, epitaxial N-doped anatase TiO₂(001) is photoactive in the visible [7]. We have used hole-mediated photodecomposition of trimethyl acetate (TMA) as a probe of photochemical activity. We find the (001)-oriented surface of epitaxial TiO_{2-x}N_x anatase ($x = \sim 0.02$) grown on Nb-doped SrTiO₃(001) by MBE is nearly as active at TMA photodecomposition with visible light as it is using broadband (visible + UV) light from a Hg arc lamp. These results reveal that at least for this particular orientation, holes are not trapped at N sites, but can freely migrate to the surface, despite the relatively long distances between substitutional N dopants. An intensive search for the mechanism behind this unusual and potentially useful result is underway.

- [2] B. O'Regan and M. Gratzel, Nature **353**, 737 (1991).
- [3] M. Gratzel, Nature **414**, 338 (2001).
- [4] M. R. Hoffmann, S. T. Martin, W. Choi, D. W. Bahnemann, Chem. Rev. **95**, 69 (1995).
- [5] O. Carp, C. L. Huisman, A. Reller, Prog. Solid State Chem. **32**, 33 (2004).
- [6] R. Asahi, T. Morikawa, T. Ohwaki, K. Aoki, Y. Taga, Science **293**, 269 (2001).
- [7] T. Ohsawa, Y Du, M.A. Henderson, V. Shutthanandan, S.A. Chambers, Phys. Rev. B **79**, 085401 (2009).



STM images of epitaxial TiO_{2-x}N_x anatase/*n*-SrTiO₃(001) with and without saturation doses of TMA, and before and after visible light irradiation: (a, a') $x=0$, clean; (b, b') $x=0$, TMA, no irradiation, where a' and b' are higher magnification images of a and b; (c) $x=0$, TMA, after visible light exposure (5.64×10^{20} photons/cm²); (d) $x=0.02$, clean; (e) $x=0.02$, TMA, no irradiation; (f) $x=0.02$, TMA, after visible light exposure (5.64×10^{20} photons/cm²). The individual TMA anions appear as bright lobes, separated by ~ 1.2 nm and ~ 1.5 nm along [010] and [100], respectively. Also shown is a structural diagram for the clean, (1 \times 4) reconstructed surface.

This work was performed in the Environmental Molecular Sciences Laboratory, a national scientific user facility sponsored by the Department of Energy's Office of Biological and Environmental Research and located at Pacific Northwest National Laboratory. This work was supported by the US Department of Energy, Office of Science, Division of Chemical Sciences.

E-mail: sa.chambers@pnl.gov

- [1] A. Fujishima and K. Honda, Nature **238**, 37 (1972).